

**Al-Karkh University of Science  
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Department of Renewable Energy Sciences  
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**Nanotechnology**

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## First lecture

### Introduction

**Nanotechnology (NT)** is an emerging multidisciplinary technique that involves application based on the synthesis of molecules in Nano-scale size range. **Nanotechnology** is also seen as new and fast emerging field that involves the manufacture, processing and application of structure, device and system by controlling shape and size in nanometer scale. The concept of nanotechnology is driven from Greek word nano ( meaning dwarf ).The Nano-particle are defined as a discrete entity that has dimensions of the order of 100 nm or less . It is the small size in combination with the chemical composition and surface that gives the nanoparticles their unique features . The alterations in physical properties serve to enhance versatility and efficacy in product development, resulting in more effective industrial and medical applications . As a result of the widespread use of nanotechnology and nano-materials , these particles find their way into the environment . Therefore , a focus on the source of nano-particles in the environment and their effects is included. This review concentrates on the recent information about analytical development. It also focuses on the applications of nanotechnology in environment, especially in the area of air and water purifications .

## History of nanotechnology

The history of nanotechnology traces the development of the concepts and experimental work falling under the broad category of nanotechnology . Although nanotechnology is a relatively recent development in scientific research , the development of its central concepts happened over a longer period of time . The emergence of nanotechnology in the 1980 was caused by the convergence experimental advances such as the invention of the scanning tunneling microscope ( STM) in 1981 and the discovery of fullerenes in 1985 , with the elucidation and popularization of a conceptual framework for the goals of nanotechnology beginning with the 1986 publication of the book Engines of Creation . The field was subject to growing public awareness and controversy in the early 2000 , with prominent debates about both its potential implications as well as the feasibility of the molecular nanotechnology , and with government moving to promote and fund research into nanotechnology . The early 2000 also saw nanotechnology , although these were limited to bulk applications of nanomaterials rather than the transformative applications envisioned by the field .

## Sources of nanoparticles in the environment

Nanoparticles are dispersed in the environment ( air , water and soil ) naturally or from unnaturally pollution sources that include industrial and combustion processes . The following table shows that .

The main sources of nanomaterials in the environment

| Source | Natural source<br>of origin   | Industrial source<br>of origin  | Engineering designed<br>nanoparticles                           |
|--------|---|---|---|
| Air    | 1 – Volcanic explosion<br>2 – Biological processes<br>3 – Nuclear processes     | 1- Burn processes<br>2 – Industrial emissions<br>3- manufacturing processes | 1 – Nanotechnology  |
| Water  | 1 – Metal sulfide nanoparticles<br>2 – Manganese oxide                          | 1 – Precipitation from<br>the atmosphere                                    | 1 – Diffusion of nanoparticle<br>durnig manufactures and<br>use |
| Soil   | 1 – Nanoscale metals<br>2 - Bio minerals<br>3 - Pools of natural organic matter | 1 - Precipitation from<br>the atmosphere                                    | 1- Diffusion of nanoparticle<br>durnig manufactures&use         |

## Difference between composite and nanomaterials

### Composite materials

Composite materials are a mixture of two or more materials, each of which has different properties from the other. When mixed with each other, we get a new material that has properties separate in terms of strength, stiffness, resistance to fracture and cracking.

Composite materials are the best alternative to metals in industrial applications. The reason for this is because they have the same strength as metals, plus, it's lighter.

Composite materials consist of two basic materials:

- **Matrix**
- **Reinforcement material**

The reinforcement material is in the form of fibers or particles that make the composite materials more resistant to cracking or fracture. The matrix works to protect reinforcement material from external damage.

Composite materials are classified according to the type of matrix into three types:

#### **1 – Polymer matrix composites:**

The matrix is this type of plastic and when the reinforcement material is glass fiber then the composite material is used in the manufacture of car glass.

#### **2 – Metal matrix composites:**

Titanium, copper and iron are used as matrix for this type of composite material.

#### **3 – Ceramic matrix composites:**

Borosilicate glass is used in some parts of aircraft engines, because of its ability to withstand high temperatures as well as its light weight.

## Nanomaterial

Nanomaterial are made of nanoparticles, whether ceramic or metals, they are much better in their properties than those of larger nanoscale counterparts, such as composite materials.

The reason for the great change in the properties of nanomaterials for composite materials is:

- 1 – Increase in proportional in area.
- 2 – Increase in chemical activity.
- 3 – Improvement in mechanical properties.
- 4- Improvement in physical properties.
- 5 - Improvement in optical properties, such as TV screens, computers, and phones.
- 6 – Increase in ability and effectiveness of magnetic properties.
- 7 – Increase in its ability to conduct electrical current.
- 8 – Improving biological properties by penetrating biological barriers and barriers.

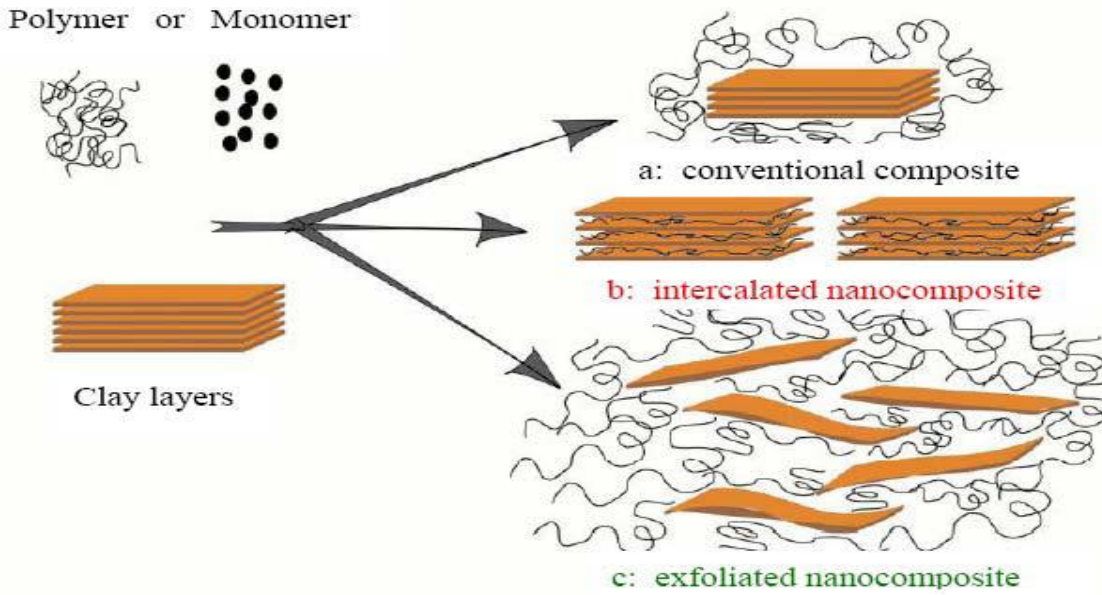
## Second lecture

### Types of composite of structure by TEM & SEM

Three main types of composites may be obtained when a **Reinforcement material** (layered clay) is associated with a **matrix** (polymer). These primarily depend on the method of preparation and the nature of components used (layered silicates, organic cation and polymer matrix)

**Transmission electron microscopy (TEM) & scan electron microscopy (SEM)** is therefore used to determine This conventional ( Traditional ) composite ( composite materials ) and nanocomposite morphology ( Intercalated & Exfoliated nanocomposites ). In addition to these two defined structures, both intercalation and partial exfoliation result in a broadening of the diffraction peak.

The following diagram and images show the types of compositions of nanoscale and composite material with a TEM & SEM .



a - conventional ( Traditional ) composite

b - Intercalated nanocomposite

c - Exofilated nanocomposite



## Using vegetable oil as a renewable energy source to obtain nanomaterials :

Plants (vegetables) considered one of the raw materials sources to obtain renewable energy through vegetables oil, and that by using them as surfactants for some materials in industry and other uses to produce renewable energy.

Among the surfactants prepared from vegetables are fatty amides ( F A ) , which organic materials that penetrate the layers of the substance that mix with it and increase the distance between the material layers ( d )and turn it into a nanocomposite. The material is test by SEM, TEM and XRD.

To calculate the distance (d) the Bragg law is used:

X-Ray diffraction study was carried out, this method is very common techniques to analyze the composition of the materials. In 1912, William Lawrence Bragg and his father William Henry Bragg formulated

### **Bragg's law $n\lambda=2d\sin\theta$**

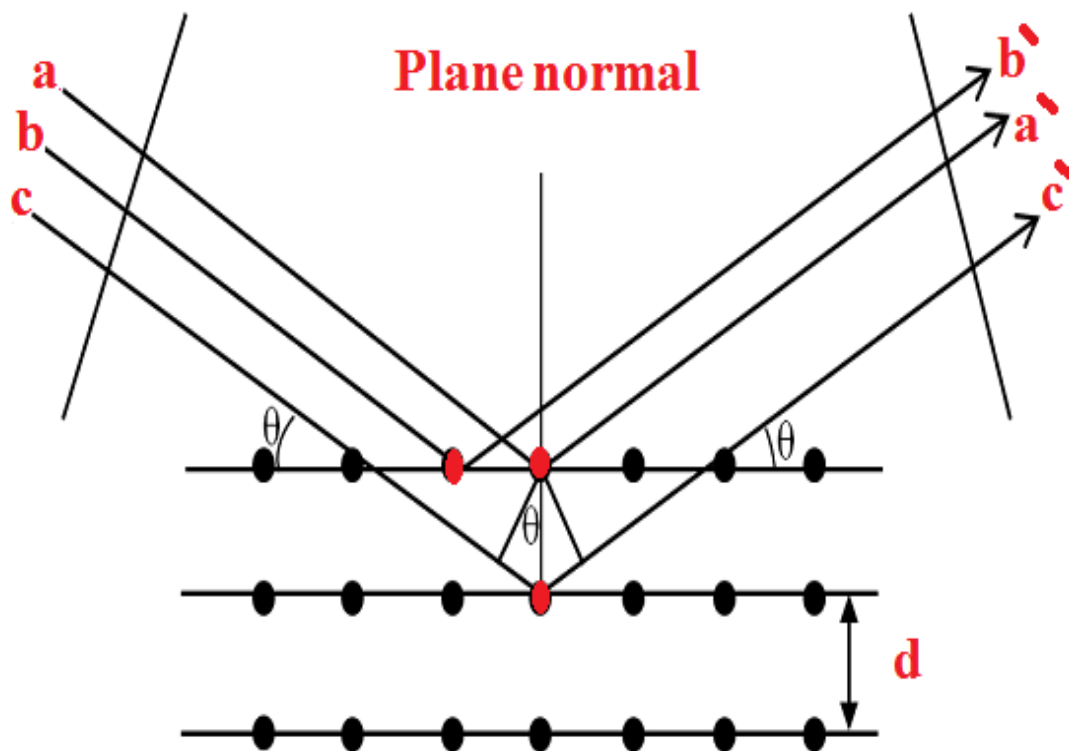
$\lambda$ = wavelength of which is = 0.15406 nm

**n**= integer represents rank diffraction peak

**d**= distance between the material layers, nm

**$\theta$** = **scattering** angle

which is the significant equation to build the relation of diffraction. When the wavelength ( $\lambda$ ) of radiation is close to the spacing ( $d$ ) between the atoms, the atoms can be considered as the diffraction grating and have the constructive and destructive interference of the scattering which is shown in Figure .



**Figure** :Clearly measured incident angle by X-ray diffraction from a layered structure

X-ray diffraction technique was used to measure the interlayer distance of the silicate layers of the clay and modified silicate layers with FA cation.

The incident beam a and b have the same incident angle and travel distance which causes the constructive interference. The incident beam c moves in phase with a and b if the extra travel distance  $2d\sin\theta$  is equal to the integral number of wavelength, or it results in destructive interference when out of phase.

The constituents of an X-Ray diffractometer are the source, sample, monochromator, detector and output device. In the hot cathode tube, the electrons are emitted from tungsten filament when heated, and travel through the electron field to impinge anode. There are two phenomena that happen when the impingement occurs. One is the inelastic collision when the partial kinetic energy of electron transfers into the continuous spectrum which is also called white radiation. Another is the characteristic spectrum which is caused by the excitation of the

inner-shell electron with collision and meanwhile, the outer-shell electron fills the inner-shell and releases energy between

the energy levels in X-Ray form. The source of the X-Ray diffractometer has a characteristic radiation, usually using  $\text{Cu K}\alpha$  and filtered by nickel as the monochromator, the wavelength of which is ( $\lambda = 0.15406 \text{ nm}$ ).

# Cement nanocomposites

## Third lecture

### Introduction

Cement consider a composite materials, is the most widely used construction materials they are rsistant to water , easily formed into vairous shapes and sizes , thr cheapest and readily available every where. One of the main types of portland cement is salt- resistant portland cement ( rpc ), the weighted percentage ( % wt ) components are in accordance with the British specifications as table shows:

Table: The main and secondary components of resistant cement rpc)  
(

| Description                                    | Formula  | % wt |
|--|--|------|
| Tricalcium silicate (C <sub>3</sub> S)         | 3CaO SiO <sub>2</sub>  | 58   |
| Dicalcium silicate (C <sub>2</sub> S)          | 2CaO SiO <sub>2</sub>  | 17   |
| Tricalcium aluminate( C <sub>3</sub> A)        | 3CaO Al <sub>2</sub> O <sub>3</sub>                                | 3    |
| Tetacalcium aluminoferrite (C <sub>4</sub> AF) | 4CaO Al <sub>2</sub> O <sub>3</sub> Fe <sub>2</sub> O <sub>3</sub> | 14   |
| Sulfur trioxide                                | SO <sub>3</sub>  | 2.56 |
| Silicon Oxide (Silica)                         | SiO <sub>2</sub>   | 21.0 |
| Ferric Oxide                                   | Fe <sub>2</sub> O <sub>3</sub>                                     | 4.65 |
| Aluminum Oxide (Alumuna)                       | Al <sub>2</sub> O <sub>3</sub>                                     | 4.07 |
| Magnesium Oxide                                | MgO  | 0.92 |
| Free lime                                      | CaO  | 63.8 |

Because the resistant cement and concrete contains pores, which leads to a weakening in the mechanical engineering properties , to contain the cement, these disadvantages were directed to applying the production of nano-cement materials can lead improvements in civil infrastructure because the mechanical strength and life of concrete structures are determined by the microstructure and the by the mass transfer in nanoscale .Some ressearchers have also studied the propertties of cement and concrete

materials with other nano-materials reinforcement , such as nano- $\text{Al}_2\text{O}_3$  , nano- $\text{ZnO}_2$ , nano- $\text{Fe}_2\text{O}_3$ , nano- $\text{CuO}$ , nano-  $\text{ZrO}_2$  and nano-clay(organo-modified-montmorillonites).

Recent studies have shown that ternary silica ( nano-silica ) particles ( $\text{NS}_t$ ) with an average diameter of about 14 nm , used as an additive to  $\text{rpc}_1$  can improve its mechanical properties , the structure , morphology and compressive strength of the synthesized nano-  $\text{rpc}_1$  were studied.  $\text{NS}_t$  was used to prepare new  $\text{rpc}_1/\text{NS}_t$  nano-composite

### **Preparation of nano- cement ( $\text{rpc}_1/\text{ternary silica}$ )**

The designed amount of  $\text{rpc}_1 / \text{NS}_t$  was prepared by an mixer on a two - roll mill , then we put the sample in an English mechanical vibrator type BIRMIHGAM , then we put the mixture in an English electric oven type GALLENHAMB at temperture of ( 950 – 1000 C ) and for a period of time up to 40 minutes and then we will process the immediately cooling, then we use a German type Sartorius sensor to weigh the samples .

The amount of  $\text{rpc}_1$  and  $\text{NS}_t$  used this study are listed in Table

**Table : The amounts of  $\text{rpc}_1$  and  $\text{NS}_t$  in the nano- cement**

| Sample identity      | Weight of $\text{rpc}_1$ , g | Weight of $\text{NS}_t$ , g |
|----------------------|------------------------------|-----------------------------|
| $\text{rpc}_1$ add 0 | 50.0                         | 0.0                         |
| $\text{rpc}_1$ add 1 | 47.5                         | 2.5                         |
| $\text{rpc}_1$ add 2 | 45.0                         | 5.0                         |
| $\text{rpc}_1$ add 3 | 42.5                         | 7.5                         |
| $\text{rpc}_1$ add 4 | 40.0                         | 10.0                        |
| $\text{rpc}_1$ add 5 | 37.5                         | 12.5                        |

add 0, add 1, add 2, add 3, add 4 and add 5 = 0, 5 , 10, 15, 20 and 25% weight of  $\text{NS}_t$  , respectively

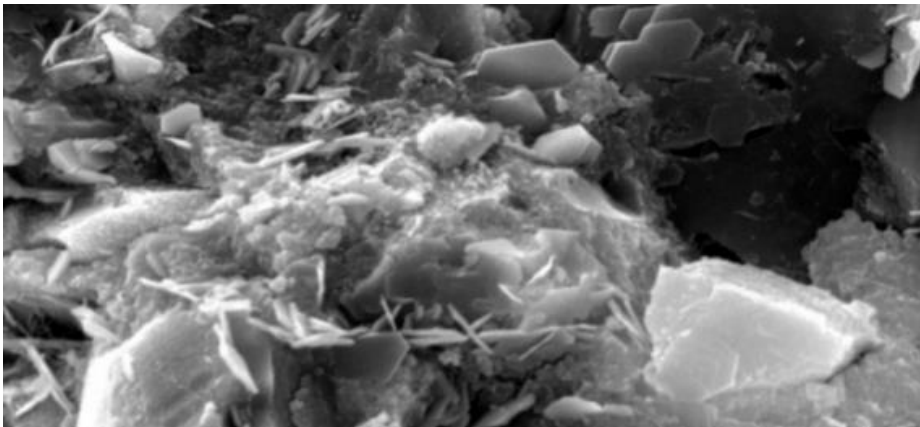
## **Characterization**

### **X-ray diffraction measurements**

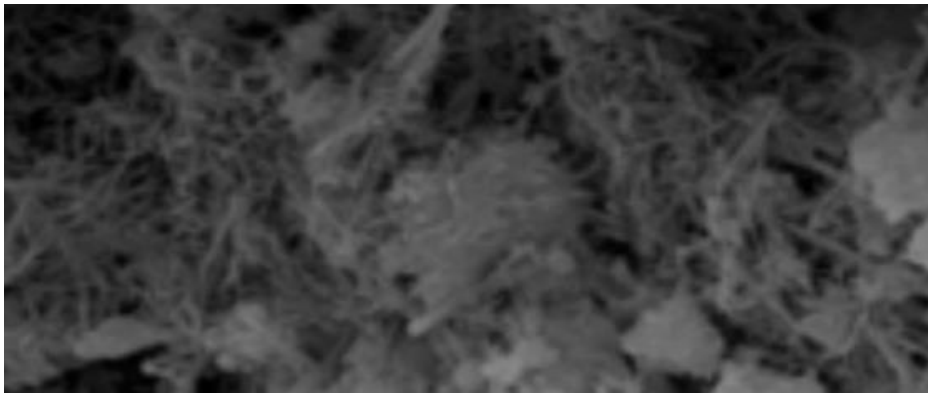
X-ray diffraction technique was used to measure the interlayer distance of the  $rpc_1$ ,  $NS_t$  and  $rpc_1 / NS_t$ . It was also used to measure the  $NS_t$  distribution of in the  $rpc_1$  matrix. Depending on the measurements of the  $2\theta$  through the XRD device, when adding 5% to 25% by weight from  $NS_t$  to  $rpc_1$  found when adding 10% gives the largest distance between the particals  $rpc_1$ .

### **Scan electron microscopy (SEM)**

SEM images of microstructure ( 5 micrometer ) are shown in figures ( 1 , 2 ) . the image in figures ( 1 ) show locally resistant portland cement (  $rpc_1$  ) , the particles appear to be conglomeratad and it has many pores between the particles . The image in figures ( 2 ) show locally resistant portland cement (  $rpc_1$  ) mixed with 10% wt from ternary silica particles ( $NS_t$ ), where it appears through the image that the pores were filled with the mixture became homogeneous nanoparticles .



**Figure 1: SEM image of microstructure of the patterns of  $rpc_1$**



**Figure 2: SEM image of microstructure of the patterns of  $rpc_1 / NS_t$**

**Conclusions, New nanocement prepared by adding ternary silica ( $NS_t$ ) to locally resistant portland cement ( $rpc_1$ ). Through the diagnosis by the XRD and SEM devices, it was found that the ratio of adding the  $NS_t$  to the  $rpc_1$  is 10 wt%. Through the diagnosis, it was observed that the percentage increased from 10 wt%, there are conglomerates in the cement mixture. This percentage leads to improve the compressive strength, other mechanical properties, a delay occurrence of fatigue and carbonization of the concrete mixture because of the occurrence of better morphology and full of pores, this shown by SEM in image 2.**

#### **Four lecture**

#### **Preparation of ternary silica particles ( Nano-silica )**

The main structure of ternary silica particle is silicon oxide ( $SiO_2$ ), where these silicates dissolve in HCl acid and in the presence of  $NH_4Cl$  that breaks down the silica gel as it is formed. Thus, silica is allowed to precipitate quickly by heating for thirty minutes on a water bath, then precipitate is filtered and washed with HCl acid and with water, it is burned in an electric oven with a degree of ( 950 – 1000 °C ) in a platinum dish for 45 minutes, then the dish is cooled and weighed, where the weight represents the sum of silica and impurities. To determine the pure silica, the silica is evaporated in the form of  $SiF_2$  using  $H_2SO_4$  acid. and the precipitate is burned with an (1150 – 1200

°C ) , then cooled and weighed . Pure silica is converted into elementary particles of microscopic by a thermal flame and the result of this process is the primary for ternary silica particles ,these particles tend to agglomerate together to give secondary particles and then agglomerate to give chains of ternary silica particles . The main reason for the strong tendency for the agglomeration of primary silica particales ( zero- dimensionalis silica ) the presence of polar hydroxyl group ( OH<sup>-</sup> ) on the surface of their particales , this leads to strong hydrogen bonds between silica particales ( the bonds are between a hydrogen atom of a a hydroxyl group in the first particle with an oxygen atom in the second particle ) so when zero ( liquid – silica ) mixtures are prepared for the puroose of their use there must be extreme heating and stirring the liquid mixture for a long period of time , then the three- dimensional structure formed breaks down and the large silica conglomerates are separated into smallar ones and figure shows that

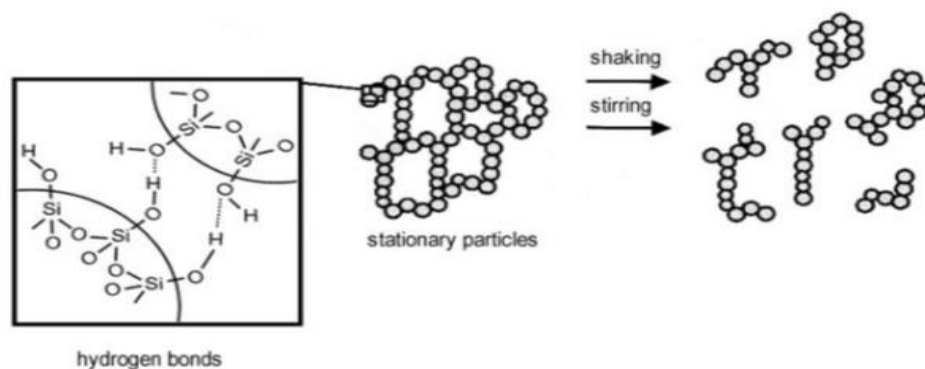


Figure: shows converting large conglomerates of triple silica into smaller conglomerates

## Methodes of preparing nanomaterials

### Introduction

Nanomaterials are nanochemical materials that are used with high quality in many industrial applications such as communications, electrons and midical filds because they contain many physical and chemical properties, and they are



prepared in several ways all in common by relying on the atomic scale, i.e. one atom towards another atom to obtain desired results, and whenever the size scale for the mass of the substance differd, the chemical activity dieeered in the sense that the smaller the scale, the higher the chemical activity of the substance. **Methodes of preparing nanomaterials :**

1 – **Physical methods**

2 – **Chemical methods**

A – Vapor- state reactions

B – Interaction in a liquid medium

3 - **Mechanical methods**

A – Mechanical composition:

B- The first monitoring and vitrification process

C- Strong diformation techniques

D- Grinding method

E - Scrubbing method

F – Electrochemical method

G- Laser so that laser beam

H – Ventilation method

## **Five lecture**

### 1 – **Physical methods**

It is prepared starting from the vapor state of the substance by heating the material or by throwing it with a beams of electron or thermal dislving it using laser beams, then the vapor is cooled by string it with a neutral gas to become more saturated and then it is placed on a cold surface quickly to avoid the occurrence of crystalline construction , then materials are prepared using nanoscale, using laser.

### 2 – **Chemical methods**

#### **A – Vapor- state reactions:**

Vaper enters, the thermal to be prepared in the CVD reactor, the material particals are mixed on a basic surface at a certain tempature and interact with other gases to form a solid strip on the surface of the base, and this method is used to prepare such as quasi- monemers.

#### **B – Interaction in a liquid medium**

Water or organic liquids are the most used, and nanomaterials are prepared by changing the chemical-physicalbalance conditions or hydrolsis to obtain spherical particales that can be

controlled in their dimensions, or through double or double chemical precipitation reactions analysis with water to obtain spherical particles whose dimensions can be controlled, techniques by using Sol gel using colloidal solutions at low temperatures.

### 3 - **Mechanical methods**

#### **A – Mechanical composition:**

By crushing a material consisting of micrometric particles from ( 1 to 30 ) into several mixtures and mixing them. Bulky materials of several tons.

#### **B- The first monitoring and vitrification process:**

By converting the atomic material into a huge piece, through two stage of melting the metal powder to form it after cooling.

#### **C- Strong deformation techniques ;**

Through by strongly deforming a crystalline material such as metal or porcelain in order to improve the hardness and ductility properties of the materials.

#### **D- Grinding method :**

It is used to produce nano-materials in powder form, where the base material is exposed to a very high energy . Grinding them with steel balls that move vibratory, planetary or vertical, and the size of the nanomaterials that are fabricated ranges from 3 to 25 nm.

#### **E - Scrubbing method :**

By placing very thin silicon strips in chemicals such as HF, and rubbing the silicon strips to obtain .

On the silicon particles slides, put these slide in a solution such as isopropanol and then in an ultrasound machine to drop the particles into solution.

#### **F – Electrochemical method :**

By placing the silicon slide at the anode and the polycarbonate slide at the negative electrode in a chemical solution , and exposing the slides to an electric.

#### **G- Laser so that laser beam :**

By exposing the material to a very high energy pulsed laser so that the laser beam interacts with the target, which leads to the volatilization of the particles of the substance and the formation of the plasma that are deposited on the and forming a thin film.

#### **H – Ventilation method :**

By exposing the material to a very low pressure vacuum thin film . And with a cold base , they are exposed to a magnetic field, which leads to the removal of the particles of the substance and depositing them in the base , forming a thin film.

### **The Tutorial of previous lectures in NT**

**Q1/** What are the main sources of nanoparticles in air (natural, industrial and engineering designed)

**Q2/** What is the difference in properties between nanoparticles and conventional composites (explain eight differences)

**Q3/** Explain with the example of using vegetable oils as renewable energy sources to obtain nanomaterials.

**Q4/** Write in detail the method of preparing the nanocement by adding ternary silica to the cement.

**Q5/** Write in detail the method of preparing the nano- ternary silica from primary silica particles ( Zero- dimensional silica).

Q6/ What are the main methods for preparing nanomaterials, have been detaild on the physical methods of preparation.

Q7/ Write in detail a method for preparing a nanomaterial so that the nanomaterials are in the form of a thin film. Q8/ Whate are the mechanical methods for preparing nanomaterials , then write in detail the method of grinding.

### Seven lecture

### Properties of nanomaterials

The properties of materials change very significatly according to their nanoscale components, so the compounds composed of stronger granules in the nanoscale , whether ceramics or minerals, are much than their counterparts in the larger size, for example about ( 10 ) times stiffer than themetal ( grain size foe example ). The is 7 mineral is the size of ordinary grains, the size of grains is measured in hundreds of nanometers, and this major in the propertes of materials in the nanoscale is caused by the following :

#### **1 – Relative increase in area:**

Nanomaterials have a larger surface area when compared to the same mass of the material produced in the larger space, and this makes the materials more chemically active and affects their strength or electrical properties.Sometimes the inert materials in the large space may be active when they are produced in the nanosphere, that is , when they are reduced.

The particles that make up matter, a very large percentage of the atoms are percent on the surface compared to those inside. Particles with a size of 30 25 nm are percent only 5 % of

them on the surface, the on surface, while others with a size of 10 nm are percent 20% of which are on the surface , and a volume of 3N , 50% of which are percent on the surface , and since the chemical reactions occurring at the surface , legal materials are more energetic than their counterparts for the most part.

## **2- Quantitative effect:**

Quantitative influences began to control the behavior of matter in the lower end scale , especially space , in the nanoscale affecting the electrical , magnetic and optical properties of materials.

## **The following are some of the properties of nanomaterials :**

### **Mechanical properties :**

The mechanical properties come on top of the properties that benefit from from the reduction in the size of the grains of the material and the presence of huge numbers of atoms on the faces of its outer surface, as the hardness of metallic materials and their alloys increases , and their resistance to facing stresses and the loads on them increases .

Formability and bearing stresses that not durability, available , and this means the creation of new types of these materials.

### **Chemical activity :**

The chemical activity of nanomaterials increases with the presence of huge numbers of atoms of substance on the faces of their external surfaces, where they act as catalysts that interact strongly with toxic gases, which filters them to play the most important role in reducing environmental pollution, and fuel cells are one of the low-cost applications of catalyst nanoparticles, and one of the important sources of new and clean energy.

### **Physical properties :**

The materials melting point values are affected by the reduction of the dimensions of its grains. The melting point of gold in its normal size reaches 1064 degrees, its weight to about 1.35 nm. 500 degrees after its grains are reduced to.

### **Optical properties :**

It is surprising and exciting that the color gold-golden yellow- changes to a transparent of natural color when its grains are reduced to less than 20 nm, and its colors change from green to orange and then red with the increase in their sizes, and this feature enables us to make high-resolution screens high contrast and clarity of colors, such as modern TV screens, computers and mobile phones.

### **Magnetic properties:**

The smaller the grains materials and the doubling of the atoms on their external surfaces, the more powerful and effective their magnetic power, which enables us to use them in large electrical generators, ship motors, the

manufacture of high – precision analysis devices, analyzers, and magnetic resonance imaging.

### **Electrical properties :**

The smallness of the materials particles sizes to less than 100 nm increases their ability to conduct electrical current, which enables us to use these materials in the micro-sensors and electronic chips.

### **Biological properties:**

Increasing the penetration ability of nanomaterials to penetrate biological barriers and membranes, and improve compatibility and biocompatibility, which facilitates the delivery of drugs and therapeutic drugs to the affected part through membranes and blood vessels.

### H.W

Q1 : What is the main property that improves the properties of nanomaterials, with the reasons being mentioned in detail.

Q2 : Chemical activity of the important properties of nanomaterials explain the following :

A – Increase the chemical activity of nanomaterials.

B – The chemical activity reduces environmental pollution .

### Eight lecture

## Nano - Mechanics

Nanomechanics : is one of the of nanotechnology, which is concerned with the branches study of mechanical properties, of elasticity, heat and motion of natural

physical systems with nanometer scales. The science of nanomechanics appeared at the meeting point of all the sciences of classical mechanics, solid state physics, and statistical mechanics, materials, and quantum chemistry. As a branch of nanotechnology, nanomechanics provides a scientific basis for nanotechnology.

Hence, the science of nanomechanics is one of the branches of nanoscience that deals with the study and application of basic mechanical properties (from elasticity, heat and movement) to physical systems of nanometer dimensions.

Nanomechanics is often reviewed as a discipline from the science of nanotechnology, and for an illustration as an applied area focusing on the mechanical properties of nanostructures.

Nanosystems are those systems with nanoscale components, nanotubes, including carbon nanotubes and boron nitride nanotubes (BNNTs).

Nanofilms, nanoparticles/nanomaterials liquids or fluids containing dispersed nanoparticles.

And as one of the basic sciences, the science of nanomechanics is based on some experimental principles (basic notes), including:

1 – Principles of general mechanics

2 – Special principles stemming from the smallness of the physical sizes for designated purposes for research study.

H.W



- And as one of the basic sciences, the science of nanomechanics is based on some experimental principles, what are the principles upon which the science of nanomechanics relied.

## Nine lecture

### Nano- chemistry

When we combine nanotechnology and chemistry, we get something called nanochemistry, which is the technique of studying and working on the smallest parts of the atom with the goal of engineering very small materials with the smallness of the number nanochemistry, and chemists and nanoparticles use a of different ways to prepare and collect small parts materials exhibit unique magnetic, the of electronic, optical, chemical and mechanical behaviors, due to its extremely small size.

Nanotechnology in chemistry represents uncommon approaches in building devices by focusing on designing on small atoms, to enjoy accurate molecular scale. This process is based on studying the synthesis and characterization of materials on the nanoscale, and focusing. How atoms interact and behave, and how they, can be controlled and manipulated into effects. And chemical reactions at the level of the atom.

Concentration of nanochemistry or nanotechnology in chemistry, also, the process of assembling single atoms in larger molecules and the resulting behavior of doing so, and this process is one of the biggest benefits of nanochemistry. As it paves the way for the creation of new materials and products, especially as it uses a wide range of chemical, physical, materials science, engineering, and biological applications.

### The role of nanotechnology in chemistry

Nanotechnology is a new concept that helps to mix chemistry with several essential factors in our life such as communications, electronics, photonics, and space engineering, transportation and medicine.

### The importance of nanochemistry

The importance of this technique stems from its great influence on our daily life and the things it can offer us, which we could not imagine a few decades ago.

We can summarize the benefits of nanotechnology in chemistry, its impact, or what you might like to call it between them via the following points:

- 1 – Create computers that are stronger, faster, and energy. Smaller in size and use less help develop batteries that will last long time production.
- 2 – Of more effective and fast medical diagnostic tools, and development of lab on chip to provide accurate and direct medical diagnosis in minutes.

3 – Nanoparticles help to better digest drugs in the human body, it also helps in facilitating the production of these drugs, and also depends on it in the production of chemotherapy drugs for specific cells such as cancer cells.

4 - Fuel efficiency in the improving cars, in addition to helping to resist corrosion, by building car parts from nanomaterials that are characterized by the fact that lighter weight, stronger strength and more resistance for chemical reactions of the metal.

5 – Nanofibers can improve resistance to fabric for stains, pollution, water and fire, without increasing the weight, thickness, or hardness of the fabric.

6 – water filter devices that are nano more than 15 to 20 nanometers wide can remove small particles as small as nanometers, and thus get rid of all bacteria and viruses that may stick to them, at low costs.

7 – Carbon nanotubes are useful in many aspects such as producing more durable and lighter activity – related equipment as tennis rackets.

8 – Most sunscreens today are made of nanoparticles, they are very effective at absorbing light even at dangerous levels of ultraviolet rays. They also spread more quickly into the skin. Factories are taking advantage of their advantage in absorbing UV rays by inserting them in composition of food packaging materials.

9 – The plastic in drinking bottles contains nano-plastics to prevent oxygen from infiltrating them and give them better carbon dioxide and moisture.

10 – Resistance programming a variety of chemical sensors that are able to identify a specific chemical with great accuracy, such as one atom out of billions. Nanosensors are useful for accurately identifying cells of the body.

